

SEC: SR-MPC:  
TIME: 3 Hrs.

GTRAND TEST-04  
EAPCET MODEL EXAM

EX. DT:  
Max. Marks: 160 M

MATHEMATICS

- The domain of  $\sin \log \left[ \frac{\sqrt{4-x^2}}{1-x} \right]$  is  
1)  $(-1, 1)$                       2)  $(-2, 1)$                       3)  $(-2, -1)$                       4)  $(1, 2)$
- If  $f$  satisfies the relation then  $f(x+y) + f(x-y) = 2f(x)f(y) \forall x, y \in R$  and  $f(0) \neq 0$  then  $f(10) - f(-10) =$   
1) 0                                      2) 1                                      3) 2                                      4) 3
- Orthocenter of an equilateral triangle ABC is the origin  $O$  if  $\overline{OA} = \bar{a}, \overline{OB} = \bar{b}, \overline{OC} = \bar{c}$  then  $\overline{AB} + 2\overline{BC} + 3\overline{CA} =$   
1)  $3\bar{c}$                                       2)  $3\bar{a}$                                       3)  $\bar{0}$                                       4)  $3\bar{b}$
- If P and Q are two points on the curve  $y = 2^{x+2}$  such that  $\overline{OP} \cdot \bar{i} = -1$  and  $\overline{OQ} \cdot \bar{i} = 2$  then the magnitude of  $|\overline{OQ} - 4\overline{OP}|$  is  
1) 10                                      2) 1                                      3) 5                                      4) 100
- If  $\overline{OA} = \bar{a}, \overline{OB} = 10\bar{a} + 2\bar{b}$  and  $\overline{OC} = \bar{b}$  where A and C are non collinear points. Let P denote the area of quadrilateral OABC and  $q$  denote the area of a parallelogram with  $\overline{OA}$  and  $\overline{OC}$  as adjacent sides then  $\frac{p}{q} =$   
1) 4                                      2) 6                                      3) 8                                      4) 10
- $\bar{b}$  and  $\bar{c}$  non collinear vectors,  $\bar{a}$  is a vector such that  $\bar{a} \cdot (\bar{b} + \bar{c}) = 4$  and  $\bar{a} \times (\bar{b} \times \bar{c}) = (x^2 - 2x + 6)\bar{b} + (\sin y)\bar{c}$  and  $0 < y \leq \pi$  then the point  $(x, y)$  is  
1)  $(1, 1)$                                       2)  $\left(1, \frac{\pi}{2}\right)$                                       3)  $(1, \pi)$                                       4)  $\left(\frac{\pi}{2}, 1\right)$
- If  $A = \begin{bmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{bmatrix}$  then  $\lim_{n \rightarrow \infty} \frac{1}{n} A^n$  is  
1) A null matrix                      2) An identity matrix                      3)  $\begin{bmatrix} 0 & 1 \\ -1 & 0 \end{bmatrix}$                       4) None
- If  $\Delta(x) = \begin{vmatrix} 1 & 1 & 1 \\ (e^x + e^{-x})^2 & (\pi^x + \pi^{-x})^2 & 2 \\ (e^x - e^{-x}) & (\pi^x - \pi^{-x}) & -2 \end{vmatrix}$  then  $\Delta(x) =$   
1)  $x^2$                                       2)  $x^2 - 1$                                       3)  $e^{x^2} - \pi^{x^2}$                                       4) 0
- If A and B are two square matrices such that  $B = -A^{-1}BA$  then  $(A+B)^2 =$   
1) 0                                      2)  $A^2 + B^2$                                       3)  $A^2 + 2AB + B^2$                                       4)  $A+B$
- By eliminating a, b, c from the homogeneous equations.  $x = \frac{a}{b-c}, y = \frac{b}{c-a}, z = \frac{c}{a-b}$  where  $a, b, c$  not all zero  
1)  $xy + yz + zx = 1$                       2)  $xy + yz + zx = -1$                       3)  $x + y + z = 0$                       4)  $x + y + z = 1$





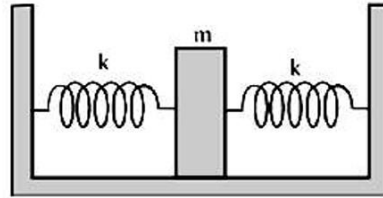
36. Three squares of a chess board are chosen at random, the probability that two are of one colour and one of another is.
- 1)  $\frac{16}{21}$                       2)  $\frac{8}{21}$                       3)  $\frac{32}{12}$                       4)  $\frac{7}{4}$
37. Five coins are tossed 3200 times. Using the poisson distribution, The approximate probability of getting five heads 2times is
- 1)  $\frac{1000}{e^{100}}$                       2)  $\frac{2500}{e^{100}}$                       3)  $\frac{5000}{e^{50}}$                       4)  $\frac{5000}{e^{100}}$
38. A box congaing 6 tickets. Two of the tickets carry a price of Rs 5/- each, the other 4 the price of Rs1. If one ticket is. drawn, the mean value of the price is.
- 1)  $\frac{14}{3}$                       2)  $\frac{7}{6}$                       3) 1                      4)  $\frac{7}{3}$
39. If an experiment with 15 observations on  $x$ , the following results were available.  
 $\sum x^2 = 2830, \sum x = 170$  one observation that was 20 was found to be wrong and was replaced by the correct value 30. Then the corrected variance is
- 1) 18.66                      2) 177.33                      3) 8.33                      4) 78.00
40. In a moderately skewed distribution the value of mean and median are 4 and 5 respectively. The value of mode in such a situation is appropriately.
- 1) 4                      2) 5                      3) 6                      4) 7
41. The equation  $\sqrt{(x-2)^2 + y^2} + \sqrt{(x+2)^2 + y^2} = 4$  represents
- 1) a pair of lines                      2) a parabola                      3) a line segment                      4) ellipse
42. The ortho-center of the triangle formed by the pair of lines  $2x^2 - xy - y^2 + x + 2y - 1 = 0$  and the line  $x + y + 1 = 0$  is
- 1)  $(-1, 0)$                       2)  $(0, -1)$                       3)  $(-1, 1)$                       4)  $(3, 5)$
43. The value of  $k > 0$  such that the angle between the lines  $4x - y + 7 = 0$  and  $kx - 5y - 9 = 0$  is  $45^\circ$  is
- 1)  $\frac{25}{3}$                       2)  $\frac{5}{3}$                       3) 3                      4) 5
44. If the axes are rotated through an angle  $45^\circ$ , and the point P has new coordinates  $(2\sqrt{2}, \sqrt{2})$ , then the original coordinates of P are
- 1)  $(1, 3)$                       2)  $(-1, 3)$                       3)  $(3, 1)$                       4)  $(3, 3)$
45.  $A = (1, -2, 3), B = (2, 1, 3), C = (4, 2, 1)$  and  $G = (-1, 3, 5)$  is the centroid of the tetrahedron  $ABCD$ .  
 If  $p = D_y$  and  $q = D_z$  then  $13p - 11q =$
- 1) 0                      2) 1                      3) -1                      4) 2
46. The point collinear with  $(1, -2, -3)$  and  $(2, 0, 0)$  among the following is
- 1)  $(0, 4, 6)$                       2)  $(0, 4, -6)$                       3)  $(0, -4, -6)$                       4)  $(0, -4, 6)$
47. For the plane  $\pi: 2x - 3y + 4z + 5 = 0$ , the points  $A = (1, 2, 0), B = (1, 2, -3)$
- 1) lie on the same side of  $\pi = 0$   
 2) lie on the opposite side of  $\pi = 0$   
 3) lie on the normal to  $\pi = 0$   
 4) none
48. The figure formed by the pairs of lines  $3x^2 + 10xy + 3y^2 = 0$  and  $3x^2 + 10xy + 3y^2 - 4x + 4y - 4 = 0$  is a
- 1) Parallelogram                      2) Rhombus                      3) Rectangle                      4) Square
49. The distance between the straight lines  $y = mx + c_1$  and  $y = mx + c_2$  is  $|c_1 - c_2|$ , when
- 1)  $m = 0$                       2)  $m = 1$                       3)  $m = 2$                       4)  $m = -2$

50. The point P is equidistant from  $A(1,3), B(-3,5)$  and  $C(5,-1)$  then  $PA =$   
 1) 5                                      2)  $5\sqrt{5}$                                       3) 25                                      4)  $5\sqrt{10}$
51. The value of 'c' in Rolle's theorem for  $f(x) = \log(x^2 + 2) - \log 3$  on  $[-1,1]$  is  
 1) 0                                      2) 1  
 3) -1                                      4) Rolle's theorem is not applicable
52. If  $x^2 + y^2 = 25$ , then  $\log_5 [\text{Max}(3x + 4y)]$  is  
 1) 2                                      2) 3                                      3) 4                                      4) 5
53. A particle moves along a line by  $S = \frac{1}{3}t^3 - 3t^2 + 8t + 5$  it changes its direction when  
 1)  $t = 1, t = 2$                                       2)  $t = 2, t = 4$                                       3)  $t = 0, t = 4$                                       4) none
54. The angle between the curves  $2y^2 = x^3$  and  $y^2 = 32x$  at  $(8,16)$  is ' $\alpha$ ' then  $\tan \alpha =$   
 1)  $\frac{3}{4}$                                       2) 2                                      3)  $\frac{1}{2}$                                       4) 1
55. At  $(2,4)$ , the curves  $y^2 = 8x$  and  $x^2 = 4y - 12$   
 1) Cut orthogonally                                      2) Touch each other  
 3) Intersect at an angle  $\frac{\pi}{3}$                                       4) Do not intersect
56. If the relative error in the radius of a circle is 0.2 then the relative error in its area is  
 1) 0.2                                      2) 0.1                                      3) 0.4                                      4) 0.3
57.  $\lim_{x \rightarrow \frac{\pi}{2}} (1 + 3 \cos x)^{\sec x} =$   
 1)  $e^2$                                       2)  $e^3$                                       3)  $e^{-2}$                                       4)  $e^{-3}$
58. If  $f(x) = \begin{cases} (1 + |\sin x|)^{\frac{a}{|\sin x|}}, & -\frac{\pi}{6} < x < 0 \\ b, & x = 0 \\ e^{\frac{\tan 2x}{\tan 3x}}, & 0 < x < \frac{\pi}{6} \end{cases}$  is Continuous at  $x = 0$  then  
 1)  $a = e^{2/3}, b = 2/3$                                       2)  $a = 2/3, b = e^{2/3}$                                       3)  $a = 1/3, b = e^{1/3}$                                       4)  $a = e^{1/3}, b = e^{1/3}$
59.  $f(x) = e^x, g(x) = \sin^{-1} x$  and  $h(x) = f(g(x))$ ; then  $\frac{h'(x)}{h(x)} =$   
 1)  $e^{\sin^{-1} x}$                                       2)  $\frac{1}{\sqrt{1-x^2}}$                                       3)  $\sin^{-1} x$                                       4)  $\frac{1}{1-x^2}$
60.  $y = a \cos 2x + b \sin 2x$ , then  $y_2 + 4y =$   
 1) 0                                      2) 1                                      3)  $y_1$                                       4)  $-y_1$
61.  $\int \frac{(\cos x)^{n-1}}{(\sin x)^{n+1}} dx =$   
 1)  $\frac{-\cot^n x}{n} + c$                                       2)  $\frac{-\cot^n x}{n+1} + c$                                       3)  $\frac{\cot^n x}{n} + c$                                       4)  $\frac{\cot^n x}{n+1} + c$
62.  $\int \frac{\sec x}{(\sec x + \tan x)^2} dx =$   
 1)  $\frac{1}{2(\sec x + \tan x)^2} + c$                                       2)  $\frac{-1}{2(\sec x + \tan x)^2} + c$   
 3)  $\frac{1}{2(\sec x + \tan x)} + c$                                       4)  $\frac{-1}{2(\sec x + \tan x)} + c$

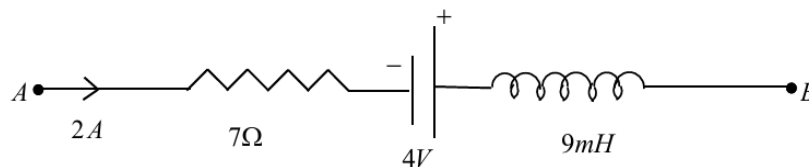
63. The anti-derivative of  $f(x) = 1 + 2^x \log 2$  is  $y = g(x)$  passes through  $\left(-1, \frac{1}{2}\right)$  then the curve  $y = g(x)$  meets the y-axis at  
 1)  $(0, -2)$                       2)  $(0, -1)$                       3)  $(0, 2)$                       4)  $(1, 0)$
64. 
$$\lim_{x \rightarrow 0} \frac{\int_0^x \sin^2 t \cos t dt}{x^3}$$
  
 1) 1                                      2)  $\frac{1}{2}$                                       3)  $\frac{1}{3}$                                       4)  $\frac{2}{3}$
65. 
$$\int_0^{\pi} \sin^6 x dx =$$
  
 1)  $\frac{5\pi}{16}$                                       2)  $\frac{35\pi}{128}$                                       3)  $\frac{5\pi}{8}$                                       4)  $\frac{5\pi}{18}$
66. 
$$\lim_{n \rightarrow \infty} \left[ \frac{1^k + 2^k + 3^k + \dots + n^k}{n^{k+1}} \right] =$$
  
 1)  $\frac{1}{k}$                                       2)  $\frac{2}{k}$                                       3)  $\frac{1}{k+1}$                                       4)  $\frac{2}{k+1}$
67. The area of the region bounded by  $x^2 = 8y, x = 4$  and the x-axis is  
 1)  $\frac{2}{3}$                                       2)  $\frac{4}{3}$                                       3)  $\frac{8}{3}$                                       4)  $\frac{10}{3}$
68. The order of the differential equation whose solution is  $y = c_1 e^x + c_2 \sin x + c_3 \cos x$  is  
 1) 4                                      2) 3                                      3) 2                                      4) 1
69. The substitution required to change  $(3y - 7x + 7)dx + (7y - 3x + 3)dy = 0$  to a homogeneous differential equation is  
 1)  $x = X + 1, y = Y$                                       2)  $x = X, y = Y + 1$   
 3)  $x = X + 1, y = Y + 1$                                       4)  $x = X + 2, y = Y + 2$
70. The solution of  $\frac{dx}{dy} + \frac{x}{y} = x^2$  is  
 1)  $\frac{1}{y} = cx - x \log x$                                       2)  $\frac{1}{x} = cy - y \log y$   
 3)  $\frac{1}{x} = cx + x \log y$                                       4)  $\frac{1}{y} = cx - y \log x$
71. If  $(-3, 2)$  lies on the circle  $x^2 + y^2 + 2gx + 2fy + c = 0$  which is concentric with the circle  $x^2 + y^2 + 6x + 8y - 5 = 0$ , then  $c =$   
 1) 11                                      2) -11                                      3) 24                                      4) -24
72. The locus of the mid point of the tangent of the circle  $x^2 + y^2 = a^2$  intercepted between the coordinate axes is  
 1)  $x^4 + y^4 = a^4$                                       2)  $x^{-2} + y^{-2} = a^{-2}$                                       3)  $x^{-2} + y^{-2} = 4a^{-2}$                                       4)  $x^4 + y^4 = 4a^4$
73. If  $(1, a), (b, 2)$  are conjugate points with respect to the circle  $x^2 + y^2 = 25$ , then  $4a + 2b = \dots\dots\dots$   
 1) 25                                      2) 50                                      3) 100                                      4) 150
74. The centre of the circle touching y-axis at  $(0, 3)$  and making an intercept 2 units on the positive x-axis is  
 1)  $(10, \sqrt{3})$                                       2)  $(\sqrt{3}, 10)$                                       3)  $(\sqrt{10}, 3)$                                       4)  $(3, \sqrt{10})$
75. If the circle  $x^2 + y^2 + 6x - 2y + k = 0$  bisects the circumference of the circle  $x^2 + y^2 + 2x - 6y + 15 = 0$  then  $k =$



89. A refrigerator with coefficient of performance  $\frac{1}{3}$  releases 200 J of heat to a hot reservoir. Then the work done on the working substance is  
 1)  $\frac{100}{3} J$                       2) 100 J                      3)  $\frac{200}{3} J$                       4) 150 J
90. At what temperature is the rms velocity of hydrogen molecule is equal to that of an oxygen molecule at  $47^\circ C$   
 1) 10 K                      2) 20 K                      3) 30 K                      4) 40 K
91. Two ideal springs of spring constants K each are attached to a block of mass m to fixed supports as shown. The time period of oscillation is



- 1)  $2\pi\sqrt{\frac{m}{K}}$                       2)  $2\pi\sqrt{\frac{m}{2K}}$                       3)  $2\pi\sqrt{\frac{2m}{K}}$                       4)  $\pi\sqrt{\frac{m}{2K}}$
92. A hollow metal sphere of radius 15cm is charged such that potential on its surface is 20 V, then the potential at the centre of sphere is  
 1) 0 V                      2) 20 V                      3) 10 V                      4) 15 V
93. In a potentiometer experiment the balancing length with a cell is 560 cm. When an external resistance of  $10\Omega$  is connected in parallel to the cell then the balancing length changes by 60cm. Find the internal resistance of a cell  
 1)  $1\Omega$                       2)  $2\Omega$                       3)  $1.2\Omega$                       4)  $2.1\Omega$
94. A particle of mass “m” and charge “q” is moving in a cyclotron with magnetic field B. The frequency of the circular motion of the particle is proportional to  
 1)  $\frac{qB}{m}$                       2)  $\frac{2m}{qB}$                       3)  $\frac{mB}{q}$                       4)  $\frac{mg}{B}$
95. A branch of circuit is shown in the figure, if current is decreasing at the rate of  $10^3 AS^{-1}$  then the potential difference between A and B is

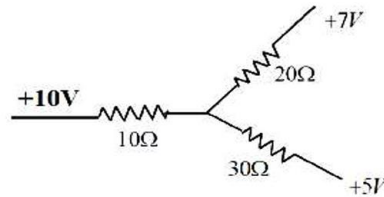


- 1) 1 V                      2) 5 V                      3) 10 V                      4) 2 V
96. Identify the logic operation performed by the following circuit
- 
- 1) OR                      2) AND                      3) NOT                      4) NAND
97. A cylindrical metallic wire is stretched to increase its length if the resistance of the wire is increased by 4% then the percentage increase in its length is  
 1) 4%                      2) 8%                      3) 1%                      4) 2%
98. The magnitude of component of the vector  $\vec{A} = 2\vec{i} + 3\vec{j}$  along the vector  $\vec{B} = (\hat{i} + \hat{j})$  is  
 1)  $\frac{5}{\sqrt{2}}$                       2)  $4\sqrt{2}$                       3)  $\frac{\sqrt{2}}{3}$                       4)  $\frac{1}{2}$

99. In the relation  $p = \frac{\alpha}{\beta} e^{\frac{\alpha z}{K\theta}}$  P is pressure, Z is distance, K is Boltzmann constant and '  $\theta$  ' is the temperature. The dimension formula of '  $\beta$  ' will be

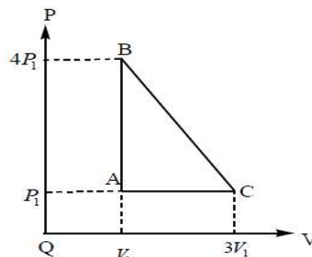
- 1)  $M^0L^2T^0$                       2)  $M^1L^2T^1$                       3)  $M^1L^0T^1$                       4)  $M^1L^0T^{-1}$

100. The current through  $10\Omega$  resistor in the figure is approximately



- 1) 0.1A                      2) 0.172 A                      3) 0.3 A                      4) 0.4 A

101. An ideal gas is taken around the cycle ABCD shown in P-V diagram. The net work done by the gas during the cycle is equal to



- 1)  $3P_1V_1$                       2)  $6P_1V_1$                       3)  $12P_1V_1$                       4)  $P_1V_1$

102. A simple pendulum is making oscillations with its bob immersed in a liquid of density  $n$  times less than the density of the bob. What is its period?

- 1)  $2\pi\sqrt{\frac{l}{ng}}$                       2)  $2\pi\sqrt{\frac{l}{\left(1-\frac{1}{n}\right)g}}$                       3)  $2\pi\sqrt{\frac{ln}{g}}$                       4)  $2\pi\sqrt{\frac{l}{(n-1)g}}$

103. Two identical glass  $\left(\mu_g = \frac{3}{2}\right)$  equiconvex lenses of focal length  $f$  are kept in contact. The space between the two lenses is filled with water  $\left(\mu_g = \frac{4}{3}\right)$ . The focal length of the combination is

- 1)  $f$                       2)  $\frac{f}{2}$                       3)  $\frac{4f}{3}$                       4)  $\frac{3f}{4}$

104. In Young's double slit experiment, the intensity on the screen at a point where path difference is  $\lambda$  is  $K$ . What will be the intensity at the point where path difference is  $\frac{\lambda}{4}$ ?

- 1)  $\frac{K}{4}$                       2)  $\frac{K}{2}$                       3)  $K$                       4) Zero

105. Two identical short bar magnets, each having magnetic moment  $M$ , are placed a distance of  $2d$  apart with axes perpendicular to each other in a horizontal plane. The magnetic induction at a point midway between them is

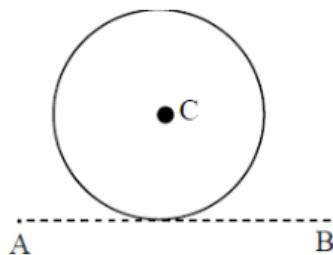
- 1)  $\frac{\mu_0}{4\pi}(\sqrt{2})\frac{M}{d^3}$                       2)  $\frac{\mu_0}{4\pi}(\sqrt{3})\frac{M}{d^3}$                       3)  $\left(\frac{2\mu_0}{\pi}\right)\frac{M}{d^3}$                       4)  $\frac{\mu_0}{4\pi}(\sqrt{5})\frac{M}{d^3}$

106. When 100 volts DC is supplied across a solenoid, a current of 1.0 ampere flows in it. When 100 volt is AC applied across the same coil, the current drops to 0.5 ampere. If the frequency of ac source is 50Hz. Then the impedance and inductance of the solenoid are

- 1)  $200\Omega$  and 0.55 henry                      2)  $100\Omega$  and 0.86 henry  
3)  $200\Omega$  and 1.40 henry                      4)  $100\Omega$  and 0.93 henry

107. The ratio of ionization energy of Bohr's hydrogen atom and Bohr's hydrogen like lithium atom is (Assume both atoms is ground state)

- 1) 1 : 1                      2) 1 : 3                      3) 1 : 9                      4) None of these
108. A wave travelling in positive X-direction with  $A = 0.2$  m has a velocity of 360 m/sec. If  $\lambda = 60$  m, then correct expression for the wave
- 1)  $y = 0.2 \sin \left[ 2\pi \left( 6t + \frac{x}{60} \right) \right]$                       2)  $y = 0.2 \sin \left[ \pi \left( 6t + \frac{x}{60} \right) \right]$
- 3)  $y = 0.2 \sin \left[ 2\pi \left( 6t - \frac{x}{60} \right) \right]$                       4)  $y = 0.2 \sin \left[ \pi \left( 6t - \frac{x}{60} \right) \right]$
109. A source and an observer move away from each other with same velocity of  $10 \text{ ms}^{-1}$  with respect to ground. If the observer finds the frequency of sound coming from the source as 1980 Hz, then actual frequency of the source is (speed of sound in air =  $340 \text{ ms}^{-1}$ )
- 1) 1950 Hz                      2) 2100 Hz                      3) 2132 Hz                      4) 2486 Hz
110. A wire of resistance  $2R$  is stretched such that its length is doubled. Then the increase in its resistance is
- 1)  $6R$                       2)  $4R$                       3)  $3R$                       4)  $2R$
111. Electro magnetic radiation of intensity  $0.6 \text{ Wm}^{-2}$  is falling on a black surface. The radiation pressure on the surface is
- 1)  $2 \times 10^{-9} \text{ Nm}^{-2}$                       2)  $3 \times 10^{-9} \text{ Nm}^{-2}$                       3)  $4 \times 10^{-9} \text{ Nm}^{-2}$                       4)  $6 \times 10^{-9} \text{ Nm}^{-2}$
112. Radiations of wavelength 400 nm incidents on a photo sensitive material of work function 2.2 eV the stopping potential is nearly
- 1) 0.1V                      2) 0.4V                      3) 0.5V                      4) 0.9V
113. Photodiodes are mostly operates in reverse biases condition because
- 1) fractional change in minority carries produces higher forward current  
 2) fractional change in majority carries produces higher reverse current  
 3) fractional change in minority carries produce higher reverse current  
 4) fractional change in majority carries produce higher forward current
114. A water drop breaks into 64 identical droplets of each surface area  $10^{-7} \text{ m}^2$ . If the surface tension of water is  $0.07 \text{ Nm}^{-1}$ , the increase in the surface energy in the process is
- 1)  $336 \times 10^{-9} \text{ J}$                       2)  $216 \times 10^{-9} \text{ J}$                       3)  $432 \times 10^{-9} \text{ J}$                       4)  $158 \times 10^{-9} \text{ J}$
115. A current ' $i$ ' is flowing through a wire of length ' $L$ '. If it is made into a circular loop of one turn, then its magnetic moment is
- 1)  $\frac{L^2 i}{4\pi}$                       2)  $\frac{L^2}{4\pi}$                       3)  $\frac{4\pi}{L^2 i}$                       4)  $4\pi L^2 i$
116. If the vertical component units of B of earth's magnetic field is  $0.5 \times 10^{-4} \text{ T}$  at a point. When aero plane of wing span 4m is moving horizontally at this place of  $360 \text{ kmh}^{-1}$  then the motional emf formed across the ends of the wings is
- 1)  $20 \times 10^{-4}$                       2)  $20 \times 10^{-2}$                       3)  $20 \times 10^{-3}$                       4)  $2 \times 10^{-4}$
117. A body of mass 2kg is sliding down on an inclined plane of inclination  $53^\circ$  to the horizontal. If coefficient of kinetic friction is 0.5, time taken by a body released from the top of an inclined plane to such the ground if the length of the inclined plane is  $10 \left( g = 10 \text{ ms}^{-2} \right), \left( \sin 53^\circ = \frac{4}{5} \right)$
- 1) 2s                      2) 4s                      3) 8s                      4) 12s
118. A thin wire of length  $l$  having linear density  $\delta$  is bent into a circular loop with C as its centre, as shown in fig. the moment of inertia of the loop about the line AB is



1)  $\frac{\delta l^3}{16\pi^2}$       2)  $\frac{\delta l^3}{8\pi^2}$       3)  $\frac{3\delta l^3}{8\pi^2}$       4)  $\frac{5\delta l^3}{16\pi^2}$

119. A and B are two satellites revolving around the earth in circular orbits with radii  $R_A$  and  $R_B$ . Their periods  $T_A$  and  $T_B$  are 8hr and 1 hr respectively. The ratio  $\frac{R_A}{R_B}$  is equal to

1)  $(8)^{\frac{1}{2}}$       2) 8      3) 4      4)  $\sqrt{8}$

120. A metal plate of area  $10^{-3} m^2$  is placed on a liquid layer of thickness 0.8mm. the coefficient of viscosity of that liquid is 24 poise. Then the horizontal force required to move the plate with a velocity of  $1 cms^{-1}$

1) 0.01N      2) 0.02N      3) 0.03N      4) 0.04N

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121. The number of orbitals associated with Quantum numbers  $n = 5$ ,  $m_s = +1/2$  is

1) 25      2) 11      3) 15      4) 50

122. The shortest wavelength of H-atom in the Lyman series is  $\lambda_1$ . The longest wavelength in the balmer series of  $He^+$  is

1)  $\frac{9\lambda_1}{5}$       2)  $\frac{5\lambda_1}{9}$       3)  $\frac{27\lambda_1}{5}$       4)  $\frac{36\lambda_1}{5}$

123. The number of significant figures in 0.00340 is \_\_\_\_\_

1) 2      2) 3      3) 5      4) 6

124. For the reaction  $Fe_2N_{(s)} + \frac{3}{2} H_{(g)} \rightleftharpoons 2Fe_s + N H_{(g)}^3$

1)  $k_c = k_p (RT)$       2)  $k_c = k_p (RT)^{-1}$       3)  $k_c = k_p (RT)^{\frac{1}{2}}$       4)  $k_c = k_p (RT)^{\frac{3}{2}}$

125. The solubility of  $Ca(OH)_2$  in water is [ Given: the solubility product of  $Ca(OH)_2$  in water =  $5.5 \times 10^{-6}$  ]

1)  $1.77 \times 10^{-6}$       2)  $1.11 \times 10^{-6}$       3)  $1.11 \times 10^{-2}$       4)  $1.77 \times 10^{-2}$

126. The ratio of number of atoms present in a simple cubic, body centered cubic and face centered cubic structure are respectively

1) 1:2:4      2) 4:2:1      3) 8:1:6      4) 4:2:3

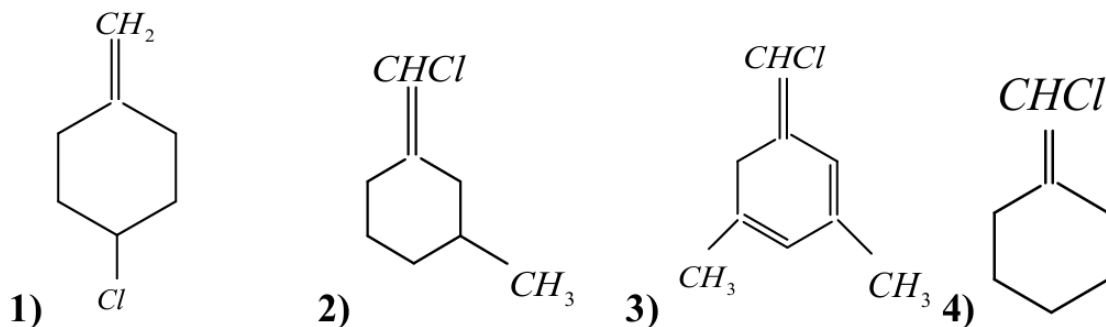
127. Freezing point of 4% aqueous solution of X is equals to freezing point of 12% aqueous solution of 'y'. If molecular weight of 'X' is A, then molecular weight of Y is \_\_\_\_\_

1) 2A      2) 3A      3) A      4) 4A

128. In Freundlich adsorption isotherm at moderate pressure the extent of adsorption (x/m) is directly proportional to  $p^x$ . The value of x is

1) Zero      2) 1      3)  $\frac{1}{n}$       4)  $\infty$

129. Among the following compounds geometrical isomerism is exhibited by



130. Excess of Isobutane on reaction with  $Br_2$  in the presence of light at  $120^{\circ} C$  gives which one of the following as the major product?



135.  $R - CN \xrightarrow[(ii)H_2O]{(i)DIBAL-H} R - Y$  consider the above reaction and identify 'Y'
- 1)  $-CONH_2$                       2)  $-CHO$                       3)  $-COOH$                       4)  $-CH_2NH_2$
136. Which among the following is not a polyester.
- 1) Glyptal                      2) PHBV                      3) Novolac                      4) Dacron
137. Which pair of oxides is Acidic in nature?
- 1)  $CaO, SiO_2$                       2)  $B_2O_3, SiO_2$                       3)  $B_2O_3, CaO$                       4)  $N_2O, BaO$
138. The total number of Isotopes of hydrogen and number of radioactive Isotopes among them respectively are
- 1) 2 and 1                      2) 3 and 2                      3) 2 and 0                      4) 3 and 1
139. Number of paramagnetic oxides among the following given oxides is \_\_\_\_\_  
 $Li_2O, CaO, Na_2O_2, KO_2, MgO$  and  $K_2O$
- 1) 1                      2) 3                      3) 0                      4) 2
140. The hybridization of the atomic orbitals of Nitrogen in  $NO_2^-$ ,  $NO_2^+$  and  $NH_4^+$  respectively are
- 1)  $sp^3, sp$  and  $sp^2$                       2)  $sp^2, sp$  and  $sp^3$                       3)  $sp^3, sp^2$  and  $sp$                       4)  $sp, sp^2$  and  $sp^3$
141. The set having ions which are coloured and paramagnetic both is
- 1)  $Cu^+, Zn^{+2}, Mn^{+4}$                       2)  $Ni^{+2}, Mn^{+7}, Hg^{+2}$                       3)  $Cu^{+2}, Cr^{+3}, Sc^+$                       4)  $Sc^{+3}, V^{+5}, Ti^{+4}$
142. The pair that has similar atomic radii is
- 1) Mn and Re                      2) Ti and Hf                      3) Sc and Ni                      4) Mo and w
143. Equal volumes of two solutions of  $p^H = 2$  and  $p^H = 5$  are mixed. The  $p^H$  of the resultant solution is
- 1) 3.5                      2) 2.5                      3) 4.7                      4) 2.3
144. Which of the following cannot be made by using Williamson's synthesis?
- 1) Methoxybenzene                      2) Benzyl p- nitrophenyl ether  
3) Methyl tertiary butyl ether                      4) Di -tert - butyl ether
145. The weight in grams of  $O_2$  formed at Pt anode during the electrolysis of aq.  $K_2SO_4$  solution during the passage of one coulomb of electricity is
- 1)  $\frac{16}{96500}$                       2)  $\frac{8}{96500}$                       3)  $\frac{32}{96500}$                       4)  $\frac{64}{96500}$
146. Molality of an aqueous solution that produces an elevation of boiling point of 1.00K at 1 atm Pressure. ( $K_b$  for water =  $0.512K \text{ kg mol}^{-1}$ )
- 1) 0.512 m                      2) 0.19 m                      3) 1.95 m                      4) 5.12 M
147. Incorrect statement about Lithium (or) its compounds is
- 1) Lithium cannot react with ethyne  
2) Lithium bicarbonate exists in solution only  
3) Lithium nitrate on heating gives Lithium nitrite and oxygen  
4) Lithium chloride exists as a dihydrate
148. The total pressure of a mixture of 8g oxygen and 14g of nitrogen contained in a 11.2 lit vessel at  $0^\circ C$  is
- 1) 0.5atm                      2) 1atm                      3) 1.5atm                      4) 2atm
149. The element having greatest difference between its first and second ionization energies, is
- 1) Ca                      2) Sc                      3) Ba                      4) K
150. One mole of an ideal gas at 300K is expanded isothermally from an initial volume of one litre to 10 liters. The  $\Delta E$  for this process is
- 1) 16.7 cal                      2) 1381 cal                      3) 9 lit atm                      4) zero
151. Which is the correct order of stabilities of the following compounds?
- 1)  $LiO_2 < NaO_2 < KO_2 < RbO_2 < CsO_2$                       2)  $LiO_2 > NaO_2 < KO_2 < RbO_2 < CsO_2$   
3)  $LiO_2 < NaO_2 < KO_2 > RbO_2 < CsO_2$                       4)  $LiO_2 > NaO_2 > KO_2 > RbO_2 > CsO_2$
152. There is no S-S bond in
- 1)  $S_2O_4^{-2}$                       2)  $S_2O_5^{-2}$                       3)  $S_2O_3^{-2}$                       4)  $S_2O_7^{-2}$
153. The crystal field stabilization energy for high spin  $d^4$  octahedral complex is

- 1)  $-0.6\Delta_0$                       2)  $-1.8\Delta_0$                       3)  $-1.6\Delta_0$                       4)  $-1.2\Delta_0$

154.  $CH_3-CH_2-Br$  on treatment with  $LiAlH_4$  gives ethane gas  $(CH_3)_3C-Br$  on same treatment gives  $H_2$  gas because

- 1) The former is  $SN_2$  and later is  $E_2$  reaction  
 2) The former is  $E_2$  and later is  $SN_2$  reaction  
 3) The former is  $SN_1$  and later is  $E_2$  reaction  
 4) The former is  $E_2$  and later is  $SN_1$  reaction

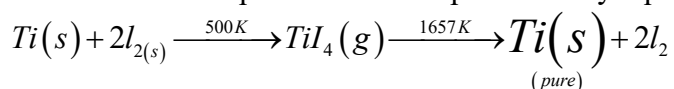
155. The increasing order of acidity among phenol, P-methyl phenol, m-nitro phenol and P-nitro phenol is

- 1) phenol < P-methyl phenol < P-nitro phenol < m-nitro phenol  
 2) P-methyl phenol < m-nitro phenol < phenol < P-nitro phenol  
 3) P-methyl phenol < phenol < m-nitro phenol < P-nitro phenol  
 4) m-nitro phenol < P-nitro phenol < phenol < P-methyl phenol

156. Paracetamol is an

- 1) Analgesic                      2) Antipyretic                      3) Both (1) & (2)                      4) None of these

157. Which method of purification is represented by equations?



- 1) Cupellation                      2) Poling                      3) Van Arkel                      4) Zone refining

158. The correct order of magnetic moments (spin only value in B.M) among the following is

- 1)  $[MnCl_4]^{2-} > [CoCl_4]^{2-} > [Fe(CN)_6]^{4-}$                       2)  $[Fe(CN)_6]^{4-} > [CoCl_4]^{2-} > [MnCl_4]^{2-}$   
 3)  $[Fe(CN)_6]^{4-} > [MnCl_4]^{2-} > [CoCl_4]^{2-}$                       4)  $[FeCl_4]^{2-} > [Fe(CN)_6]^{4-} > [CoCl_4]^{2-}$

159. **Column I**

- (A)  $[Ni(NH_3)_6]^{2+}$   
 (B)  $[Co(NO_2)_6]^{3-}$   
 (C)  $[Cu(CN)_4]^{3-}$   
 (D)  $[Pt(NH_3)_2Cl_2]$

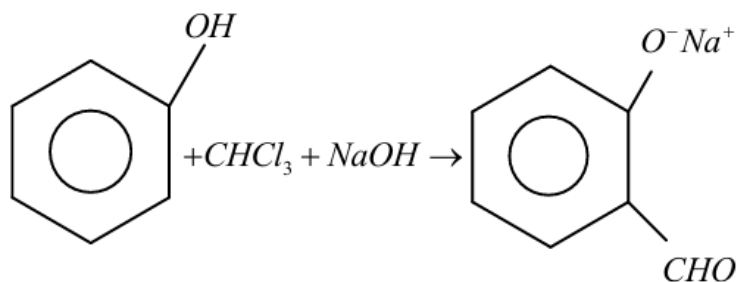
- 1) A-p; B-s; C-r; D-q  
 3) A-s; B-q; C-p; D-r

**Column II**

- (p)  $sp^3$  - hybridization  
 (q)  $d^2sp^3$  - hybridization  
 (r)  $dsp^2$  - hybridization  
 (s)  $sp^3d^2$  - hybridization

- 2) A-q; B-p; C-s; D-r  
 4) A-r; B-s; C-q; D-p

160.



The electrophile involved in the above reaction is

- 1) Trichloromethyl anion  $(\bar{C}Cl_3)$                       2) Formyl cation  $(\oplus CHO)$   
 3) Dichloromethyl cation  $(\oplus CHCl_2)$                       4) Dichlorocarbene  $(:CCl_2)$